

Towards the Synthesis of Spongistatin 1: Diastereoselective Synthesis of the C(36)-C(45) Subunit

Glenn C. Micalizio and William R. Roush1*

Department of Chemistry, University of Michigan, Ann Arbor, MI 48109

Received 9 February 1999; accepted 17 February 1999

Abstract: An efficient and highly diastereoselective 12 step synthesis of the C(36)-C(45) subunit (2a) of spongistatin 1 is described. The synthesis features highly diastereoselective α -alkoxyallylation reactions using the γ -alkoxy substituted allylstannanes 5 and 6, and a thermodynamically controlled intramolecular Michael addition to close the pyran. © 1999 Elsevier Science Ltd. All rights reserved.

The spongistatins and altohyrtins are a structurally complex class of sponge-derived macrolide polyethers possessing unparalleled inhibitory activity against a subset of chemoresistant tumor types.² Spongistatin 1 (1, also known as altohyrtin A) is among the most active members of this family.³ The extremely meager natural supply, novel structural features, and potent biological activities have defined the spongistatins and altohyrtins as interesting targets for total synthesis. Elegant total syntheses of spongistatins 2 and 1 have been reported by the Evans⁴ and Kishi⁵ groups, and additional studies directed towards the synthesis of this family have appeared.⁶ We report herein a very brief and highly diastereoselective synthesis of the spongistatin C(36)-C(45) fragment, represented by the F ring pyran unit 2a.

Our strategy from the outset called for 2a to be assembled by the intramolecular Michael cyclization of 3 (R = H), which in turn would be assembled by a series of diastereoselective γ -alkoxyallylstannylations⁷ of a suitable synthetic equivalent of methyl malondialdehyde, 4. During the course of this work we developed the (Z)- γ -alkoxymethallylstannane reagent 5^8 for introduction of the C(38)-C(39) diol unit in a fully differentiated manner. The γ -alkoxyallyl stannane 6^9 (PMP = p-methoxyphenyl) similarly proved useful for introducing the differentiated C(41) and C(42) diol unit of 3.

Chelate controlled addition of the allylstannane reagent 5 to the readily available aldehyde 7^{10} (MgBr₂•Et₂O, CH₂Cl₂, -78 °C) provided the homoallylic alcohol 8a in 93% yield. The diastereoselectivity of this reaction was excellent ($\geq 20:1$ d.s.). Treatment of 8a with triethylsilyl triflate (TES-OTf) provided the corresponding TES ether in 92% yield. We initially transformed 8a into aldehyde 9, which we imagined

would be a suitable substrate for a Lewis acid promoted α -alkoxyallylation with 6. However, this reaction gave a complex mixture of products lacking disubstituted olefin units, and we suspected that 9 may have undergone an intramolecular Prins reaction. Accordingly, we decided to chemically differentiate the C(37)-olefin before proceeding with the synthesis. This was accomplished by asymmetric dihydroxylation of TES

ether **8b**, which provided an 8:1 mixture of diastereomeric diols in 88% yield. ¹² Treatment of the resulting diol with triphosgene and pyridine in CH₂Cl₂ provided carbonate **10** in 99% yield. The major diastereomer was assigned the stereochemistry shown by application of Sharpless' empirical mnemonic device. ^{12,13} Removal of the PMB ether by treatment of **10** with DDQ¹⁴ in wet CH₂Cl₂ provided the primary alcohol which was oxidized to the aldehyde using the standard Swern protocol. ¹⁵ The resulting 2,3-anti aldehyde **11** was then treated with allylstannane reagent **6** and BF₃•Et₂O in CH₂Cl₂ at -78 °C, which provided the desired alcohol **12** with >20:1 diastereoselectivity (93% yield). The terminal olefin was oxidatively cleaved via a two step procedure (OsO₄, NMO followed by NaIO₄, 82%). The crude aldehyde was then subjected to a Horner-Wadsworth-Emmons reaction using (EtO)₂POCH₂CO₂Me, LiCl and DBU in acetonitrile. ¹⁶ When the HWE reaction was performed using excess LiCl at room temperature, the TES group migrated from the C(39) hydroxyl to the C(41) hydroxyl, and the C(39) hydroxyl subsequently underwent 1,4-addition to the newly introduced enoate. This one pot olefination, silyl migration, 1,4-addition reaction sequence furnished the targeted pyran system as a 10:1 mixture of diastereomers in 68% yield. The stereochemistry of the diastereomeric pyrans was assigned by ¹H NMR studies, which showed that the major pyran **13** possessed the incorrect (axial) stereochemistry at C(43).

Initial attempts to equilibrate the mixture of pyran diastereomers were unsuccessful (BnMe₃N⁺-OMe, MeOH, THF, O °C; KOt-Bu, THF, 0 ° to 23 °C; DBU, DMF, 100 °C); in general, pyran 13 was recovered without significant equilibration to 2b. We speculated that the inability to equilibrate this mixture was due to a remote steric effect of the C(41)-triethylsilyl ether, which causes the C(42) p-methoxyphenyl ether to adopt a

conformation anti to the C(41)-C(42) bond. As long as the C(43) substituent is axial, the C(42) aryl ether then occupies an unhindered quadrant, devoid of any significant gauche interactions.¹⁷ However, if the C(43) side chain is equatorial, then the C(42) aryl ether experiences significant gauche interactions with either with C(44), or with the C(41)-TES ether. Therefore, in order to relieve these interactions, the mixture of 13 and 2b was treated with PPTs in methanol, which provided the corresponding 10:1 mixture of C(41)-alcohols 14 and 2a (93%). Fortunately, treatment of this mixture with DBU in DMF at 80° C furnished an equilibrated 9:1 mixture of the desired pyran 2a (now major) and the C(43) axial epimer 14 in 56% yield.

Armed with the insight provided by these equilibration studies, the synthesis of 2a was improved as follows. When the Horner-Wadsworth-Emmons olefination of 15 (generated from 12 as described above) was performed using 3 equiv of (EtO)₂P(O)CH₂CO₂Me, 3 equiv of LiCl and 3 equiv of DBU in CH₃CN at -35 ° to -10 °C, the corresponding enoate was isolated in 88% yield. Subsequent deprotection of the TES ether via exposure to PPTs in methanol then gave diol 16 in 94% yield. Finally, the 1,4-addition was accomplished by treatment of 16 with DBU under thermodynamic conditions (DMF, 95 °C, 22 h), from which the desired pyran 2a was obtained with 25: 1 diastereoselectivity in 65% yield.

In summary, we have developed a 12 step synthesis of the C(36) to C(45) subunit (2a) of spongistatin 1 that proceeds in 28% yield from the known aldehyde 7. The synthesis features highly diastereoselective α -alkoxyallylation reactions using the γ -alkoxy substituted allylstannanes 5 and 6, and a thermodynamically controlled intramolecular Michael addition to close the pyran ring system. Additional progress towards the synthesis of spongistatin 1 will be reported in due course.

Acknowledgment. Financial support provided by the National Institutes of Health (GM 38436) is gratefully acknowledged.

References

- (1) Address correspondence to this author at the Department of Chemistry, University of Michigan, Ann Arbor, MI 48109-1055; e-mail: roush@umich.edu.
- (2) (a) Pettit, G. R.; Cichacz, Z. A.; Herald, C. L.; Gao, F.; Boyd, M. R.; Schmidt, J. M.; Hamel, E.; Bai, R. J. Chem. Soc., Chem. Commun. 1994, 1605. (b) Bai, R.; Taylor, G. F.; Cichacz, Z. A.; Herald, C. L.; Kepler, J. A.; Pettit, G. R.; Hamel, E. Biochemistry 1995, 34, 9714.
- (3) (a) Pettit, G. R.; Cichacz, Z. A.; Gao, F.; Herald, C. L.; Boyd, M. R.; Schmidt, J. M.; Hooper, J. N. A. J. Org. Chem. 1993, 58, 1302. (b) Kobayashi, M.; Aoki, S.; Sakai, H.; Kawazoe, K.; Kihara, N.; Sasaki, T.; Kitagawa, I. Tetrahedron Lett. 1993, 34, 2795. (c) Kobayashi, M.; Aoki, S.;

- Kitagawa, I. Tetrahedron Lett. 1994, 1243. (d) Kobayashi, M; Aoki, S.; Sakai, H.; Kihara, N.; Sasaki, T.; Kitagawa, I. Chem. Pharm. Bull. 1993, 41, 989.
- (4) (a) Evans, D. A.; Coleman, P. A.; Dias, L. C. Angew. Chem. Int. Ed. 1997, 36, 2738. (b) Evans, D. A.; Trotter, B. W.; Côté, B.; Coleman, P. J. Angew. Chem. Int. Ed. 1997, 36, 2741. (c) Evans, D. A.; Trotter, B. W.; Côté, B.; Coleman, P. J.; Dias, L. C.; Tyler, A. N. Angew. Chem. Int. Ed. 1997, 36, 2744.
- (5) (a) Guo, J.; Duffy, K. J.; Stevens, K. L.; Dalko, P. I.; Roth, R. M.; Hayward, M. M.; Kishi, Y.
 Angew. Chem. Int. Ed. 1998, 37, 187. (b) Hayward, M. M.; Roth, R. M.; Duffy, K. J.; Dalko, P.
 I.; Stevens, K. L.; Guo, J.; Kishi, Y. Angew. Chem. Int. Ed. 1998, 37, 192.
- (6) (a) Claffey, M. M.; Heathcock, C. H. J. Org. Chem. 1996, 61, 7646. (b) Hayes, C. H.; Heathcock, C. H. J. Org. Chem. 1997, 62, 2678. (c) Paterson, I.; Oballa, R. M.; Norcross, R. D. Tetrahedron Lett. 1996, 37, 8581. (d) Paterson, I.; Keown, L. E. Tetrahedron Lett. 1997, 38, 5727. (e) Paterson, I.; Oballa, R. M. Tetrahedron Lett. 1997, 38, 8241. (f) Paterson, I.; Wallace, D. J.; Gibson, K. R. Tetrahedron Lett. 1997, 38, 8911. (g) Paquette, L. A.; Zuev, D. Tetrahedron Lett. 1997, 38, 5115. (h) Paquette, L. A.; Braun, A. Tetrahedron Lett. 1997, 38, 5119. (i) Smith, A. B.; Zhuang, L.; Brook, C. S.; Boldi, A. M.; McBriar, M. D.; Moser, W. H.; Murase, N.; Nakayama, K.; Verhoest, P. R.; Lin, Q. Tetrahedron Lett. 1997, 38, 8667. (j) Smith, A. B.; Zhuang, L.; Brook C. S.; Lin, Q.; Moser, W. H.; Trout, R. E. L.; Boldi, A. M. Tetrahedron Lett. 1997, 38, 8671. (k) Smith, A. B.; Lin, Q.; Nakayama, K.; Boldi, A. M.; Brook, C. S.; McBriar, M. D.; Moser, W. H.; Sobukawa, M.; Zhuang, L. Tetrahedron Lett. 1997, 38, 8675. (1) Lemaire-Audoire, S.; Vogel, P. Tetrahedron Lett. 1998, 39, 1345. (m) Hermitage, S. A.; Roberts, S. M. Tetrahedron Lett. 1998, 39, 3563. (n) Hermitage, S. A.; Roberts, S. M.; Watson, D. J. Tetrahedron Lett. 1998, 39, 3567. (o) Terauchi, T.; Nakata, M. Tetrahedron Lett. 1998, 39, 3795. (p) Zemribo, R.; Mead, K. T. Tetrahedron Lett. 1998, 39, 3895. (q) Smith, A. B.; Lin, Q. Bioorganic & Medicinal Chemistry Letters 1998, 8, 567. (r) Crimmins, M. T.; Washburn, D. G. Tetrahedron Lett. 1998, 39, 7487. (s) Paterson, I.; Wallace, D. J.; Oballa, R. M. Tetrahedron Lett. 1998, 39, 8545.
- (7) (a) Keck, G. E.; Abbott, D. E.; Wiley, M. R. Tetrahedron Lett. 1987, 28, 139. (b) Yamamoto, Y.;
 Asao, N. Chem. Rev. 1993, 93, 2207. (c) Marshall, J. A. Chem. Rev. 1996, 96, 31.
- (8) (Z)- γ -(tert-Butyldimethylsilyloxy)methallyltributylstannane (5) was prepared in 64% yield by metallation of tert-butyldimethylsilyl methallyl ether followed by addition of Bu₃SnCl.

(9) (Z)-γ-Alkoxyallylstannane 6 was prepared via alkylation of p-methoxyphenol with allyl bromide followed by metallation with s-BuLi and quenching with Bu₃SnCl.

- (10) Smith, A. B., III; Qiu, Y.; Jones, D. R.; Kobayashi, K. J. Am. Chem. Soc. 1995, 117, 12011.
- (11) Reetz, M. T. Angew. Chem. Int. Ed. Engl. 1984, 23, 556.
- (12) Kolb, H. C.; VanNieuwenhze, M. S.; Sharpless, K. B. Chem. Rev. 1994, 94, 2483.
- (13) The stereochemistry of this diol is of no consequence to the synthesis, since ultimately it will be oxidatively cleaved to give a methyl ketone needed for assembly of the E ring. Therefore, the new stereocenter generated in this reaction was not rigorously assigned. The asymmetric dihydroxylation protocol was employed since use of OsO₄ and NMO (THF, acetone, pH 7 buffer) gave a ca. 70:30 mixture of the diastereomeric diols.
- (14) Horita, K.; Yoshioka, T.; Tanaka, T.; Oikawa, Y.; Yonemitsu, O. Tetrahedron 1986, 42, 3021.
- (15) Tidwell, T. T. Synthesis 1990, 857.
- (16) Blanchette, M. A.; Choy, W.; Davis, J. T.; Essenfeld, A. P.; Masamune, S.; Roush, W. R.; Sakai, T. Tetrahedron Lett. 1984, 25, 2183.
- (17) Hoffmann, R. W. Angew. Chem., Int. Ed. Engl. 1992, 31, 1124.